## Medical Calculations and Conversions

## Most of the material courtesy of Elizabeth Warren, RVT (modified from VSPN)

## Terms:

Kilo $=(k)=1000=10^{3}$
Basic Unit $=1=10^{0}$
Deci $=(\mathrm{d})=0.1=10^{-1}$
Centi $=(\mathrm{c})=0.01=10^{-2}$
Milli $(\mathrm{m})=0.001=10^{-3}$
Micro $(\mathrm{m})=0.000001=10^{-6}$
Pico $(p)=0.000000000001=10^{-12}$

More commonly used units:
1000 milliliters ( ml ) $=1.0$ liter ( L )
1000 microliters (ul) $=1.0 \mathrm{ml}$
$1,000,000 \mathrm{ul}=1.0 \mathrm{~L}$
$30 \mathrm{ml}=1.0$ ounce (oz)
$5 \mathrm{ml}=1.0$ teaspoon (tsp)
$8 o z=1$ cup
$\%$ solution $=$ grams $/ 100 \mathrm{ml}$

> 1000 grams $(\mathrm{g})=1.0$ kilograms $(\mathrm{kg})$ 1000 milligrams $(\mathrm{mg})=1.0$ gram 1000 micrograms $(\mathrm{ug})=1.0 \mathrm{mg}$ 2.2 pounds $(\#, \mathrm{lb})=1.0 \mathrm{~kg}$ 1.0 grain $(\mathrm{gr})=64.8 \mathrm{mg}{ }^{* *}$ This is really only relevant these days for Phenobarbital. This drug is one of the few still described in grains rather than milligrams

## Weight (gram, pound, ounce)

| 1 grain (gr) $=0.0648$ grams (g) |
| :---: |
| 1 grain $(\mathrm{gr})=64.8$ milligrams $(\mathrm{mg})$, sometimes 60 mg is used |
| 1 gram (g) = 0.001 kilograms (g) |
| 1 gram (g) = 1,000,000 micrograms (ug) |
| $1 \mathrm{gram}(\mathrm{g})=1,000,000,000,000$ picograms (pg) |
| 1 gram (g) = 1000 milligrams (mg) |
| 1 gram (g) = 15 grain (gr) |
| 1 kilogram (kg) = 1000 grams (g) |
| 1 kilogram (kg) = 2.2 pounds (lb or \#) |
| $1 \mathrm{micrograms} / \mathrm{gram}$ (ug/g) = 1 parts per million (ppm) |
| 1 milligram (mg) = 1000 micrograms (ug) |
| 1 milligram/kilogram ( $\mathrm{mg} / \mathrm{kg}$ ) = 1 parts per million (ppm) |
| 1 ounce (oz) = 28.4 grams (g) |
| 1 pound (lb or \#) = 0.45 kilograms (kg) |
| 1 pound (lb or \#) = 16 ounces (oz) |
| 1 ton = 1016 kilograms (kg) |
| 1 ton = 2000 pounds (lb or \#) |

Length (meter, inch, foot)

| 1 centimeter (cm) $=0.39$ inches (in) |
| :--- |
| 1 foot (ft) $=12$ inches (in) |
| 1 foot (ft) $=30.5$ centimeters (cm) |
| 1 inch (in) $=2.54$ centimeters (cm) |
| 1 mile $=1.6$ kilometers (km) |
| 1 mile $=1760$ yards (yd) |
| 1 mile $=5280$ feet ( ft$)$ |
| 1 yard (yd) $=0.9$ meters (m) |
| 1 yard (yd) $=3$ feet (ft) |

## Dilutions

| $1: \mathrm{X}$ | $\mathrm{mg} / \mathrm{mL}$ | $\%$ |
| :--- | :--- | :--- |
| $1: 10$ | $100 \mathrm{mg} / \mathrm{mL}$ | $10 \%$ |
| $1: 100$ | $10 \mathrm{mg} / \mathrm{mL}$ | $1 \%$ |
| $1: 1000$ | $1 \mathrm{mg} / \mathrm{mL}$ | $0.1 \%$ |
| $1: 10,000$ | $0.1 \mathrm{mg} / \mathrm{mL}$ | $0.01 \%$ |
| $1: 100,000$ | $0.01 \mathrm{mg} / \mathrm{mL}$ | $0.001 \%$ |

Body surface area:

| kg | m2 | kg | m2 | kg | m2 | kg | m2 | kg | m2 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 0.5 | 0.06 |  |  |  |  |  |  |  |  |
| 1.0 | 0.10 | 11.0 | 0.49 | 21.0 | 0.76 | 31.0 | 0.99 | 41.0 | 1.19 |
| 2.0 | 0.15 | 12.0 | 0.52 | 22.0 | 0.79 | 32.0 | 1.01 | 42.0 | 1.21 |
| 3.0 | 0.20 | 13.0 | 0.55 | 23.0 | 0.81 | 33.0 | 1.03 | 43.0 | 1.23 |
| 4.0 | 0.25 | 14.0 | 0.58 | 24.0 | 0.83 | 34.0 | 1.05 | 44.0 | 1.25 |
| 5.0 | 0.29 | 15.0 | 0.60 | 25.0 | 0.85 | 35.0 | 1.07 | 45.0 | 1.26 |
| 6.0 | 0.33 | 16.0 | 0.63 | 26.0 | 0.88 | 36.0 | 1.09 | 46.0 | 1.28 |
| 7.0 | 0.36 | 17.0 | 0.66 | 27.0 | 0.90 | 37.0 | 1.11 | 47.0 | 1.30 |
| 8.0 | 0.40 | 18.0 | 0.69 | 28.0 | 0.92 | 38.0 | 1.13 | 48.0 | 1.32 |
| 9.0 | 0.43 | 19.0 | 0.71 | 29.0 | 0.94 | 39.0 | 1.15 | 49.0 | 1.34 |
| 10.0 | 0.46 | 20.0 | 0.74 | 30.0 | 0.96 | 40.0 | 1.17 | 50.0 | 1.36 |

Volume (fluid ounce, liter)

| 1 cup (c) = 240 milliliters (mL) | 1 milliliter (mL) $=1000$ microliters (mL) |
| :--- | :--- |
| 1 cup (c) $=8$ ounces (fl oz ) | 1 ounce (fl oz) $=30$ milliliters (mL) |
| 15 drops (ggt) $=1$ milliliter (mL) | 1 pint (pt) $=0.47$ liters (L) |
| 1 gallon (gal) $=3.79$ liters (L) | 1 pint (pt) $=2$ cups (c) |
| 1 gallon (gal) $=4$ quarts (qt) | 1 quart (qt) $=0.95$ liters (L) |
| 1 liter (L) = 1000 milliliters (mL) | 1 quart (qt) $=2$ pints (pt) |
| 1 milliliter (mL) $=0.34$ ounces (fl oz) | 1 tablespoon (tbs) $=15$ milliliters (mL) |
| 1 milliliter (mL) $=1$ cc (centimeter cubic) | 1 teaspoon (tsp) $=5$ milliliters (mL) |

## Temperature

${ }^{\circ} \mathrm{F}=\left({ }^{\circ} \mathrm{C} \times 9 / 5\right)+32$
${ }^{\circ} \mathrm{C}=\left({ }^{\circ} \mathrm{F}-32\right) \times 5 / 9$
$[9 / 5=1.8]$
[5/9 = 0.555 etc.]

Approximate Daily Energy (kcal) and fluid ( mL ) requirements of dogs (in kg):

| kg | mL/kcal | kg | mL/kcal | kg | mL/kcal | kg | mL/kcal |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 132 | 10 | 742 | 19 | 1201 | 28 | 1607 |
| 2 | 222 | 11 | 797 | 20 | 1248 | 29 | 1650 |
| 3 | 301 | 12 | 851 | 21 | 1295 | 30 | 1692 |
| 4 | 373 | 13 | 904 | 22 | 1341 | 35 | 1899 |
| 5 | 441 | 14 | 955 | 23 | 1386 | 40 | 2100 |
| 6 | 506 | 15 | 1006 | 24 | 1431 | 45 | 2293 |
| 7 | 568 | 16 | 1056 | 25 | 1476 | 50 | 2482 |
| 8 | 628 | 17 | 1105 | 26 | 1520 |  |  |
| 9 | 686 | 18 | 1154 | 27 | 1564 |  |  |

Approximate Daily Energy (kcal) and fluid ( mL ) requirements of cats:

| $\mathbf{k g}$ | $\mathrm{mL} / \mathrm{kcal}$ | $\mathbf{k g}$ | $\mathrm{mL} / \mathrm{kcal}$ | $\mathbf{k g}$ | $\mathrm{mL} / \mathrm{kcal}$ |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 1.0 | 80.0 | 2.5 | 159.1 | 4.0 | 226.3 |
| 1.5 | 108.4 | 3.0 | 182.4 | 4.5 | 247.2 |
| 2.0 | 134.5 | 3.5 | 204.7 | 5.0 | 267.5 |

## Other Equations and Calculations

* Molarity (M) = \# of moles of solute / liters (L) of solution
* Milliliters (mL) x M (molarity) = \# of millimoles
* Normality ( N ) = \# of gram equivalents solute / L
* Milliequivalent $(\mathrm{mEq})=$ mmoles x valence $=$ ( mg x valence) / mw (molecular weight)
*. 1 gram ( g ) $=1,000,000,000,000$ picograms ( pg )
* Serum osmolality $=2(\mathrm{Na}+\mathrm{K})+\frac{\mathrm{BUN}}{28}+\frac{\mathrm{Glu}}{18}$
* Anion gap $=(\mathrm{Na}+\mathrm{K})-(\mathrm{Cl}+\mathrm{HCO} 3)$
* Globulin = Total Protein - Albumin
- A/G ratio = Albumin/Globulin
*. Corrected Calcium = Calcium - Albumin + 3.5
* Sodium/Potassium ratio $=\mathrm{Na} / \mathrm{K}$
* French to English measurement of tubular instruments: $1 \mathrm{Fr}=0.33 \mathrm{~mm}$
* Subcutaneous fluid dose» $60 \mathrm{~mL} / \mathrm{kg}$
* RER (Resting Energy Requirement) $=70 \mathrm{x}$ weight ( kg ) to the 0.75 power
* MER (Maintenance Energy Requirement) = activity or illness factor $x$ RER
* Food dosage = kcal required/caloric density of food
* whole blood transfusion mL needed

DOG $=(\underline{\text { desired PCV - current PCV }}) \times(\mathrm{kg}) \times 80$
PCV of donor blood
$C A T=($ desired PCV - current PCV $) \times(\mathrm{kg}) \times 70$
PCV of donor blood

## Problem Sets - common mathematical calculations Courtesy of Rose Peters, DVM, DACVIM (Neurology/ Neurosurgery)

Note: This primer assumes you have a basic understanding of mathematical principles (addition, subtraction, multiplication, division, decimal system, and fractions). Please go to www.math.com for a reminder of basic mathematical principles if needed. For most of my medical calculations, I like to use the method of "unit cancellation". I will demonstrate this with each problem presented below as it applies. This method will keep you on the right track in terms of making sure you are making the right choice to multiply versus divide to make your calculations. It keeps you from having to memorize lots of equations and "short-cuts" and is a good way to double check yourself.

Here is an example with explanation of how this might work for us. We can convert $10 \mathrm{mg} / \mathrm{lb}$ to $\mathrm{g} / \mathrm{kg}$ in the following way:
We know that there are 2.2 lb in each kilogram, so $10 \times 2.2=22 \mathrm{mg} / \mathrm{kg}$
How do we know that it is $10 \times 2.2$ and not 10/2.2?? If we use unit cancellation, you will be correct every time!
They key is to follow a system that requires putting each number into a "fraction" format:
Step 1 - Put each number into a fraction and make sure each fraction is in the correct orientation
Step 2 - Multiply across the fractions (this depends on you knowing how to multiply fractions)
Step 3 - Cancel any units that are found both on top (in the numerator) and bottom (denominator) of the final answer
Step 4 - Do your units match the desired result? Does your answer make sense?

Using the problem above, we will set up our fractions as follows:
$\underline{10 \mathrm{mg}} \times \underline{2.2 \mathrm{lb}}=\underline{10 \mathrm{mg}} \times \underline{2.2 \mathrm{bb}}=10 \times 2.2=\underline{22 \mathrm{mg}}$ This has given us part of our answer. Now we must convert to $\mathrm{g} / \mathrm{kg}$
$\mathrm{lb} \quad \mathrm{kg} \quad \mathrm{Hb} \quad \mathrm{kg} \quad \mathrm{kg}$
**Notice how the conversion of $2.21 \mathrm{~b} / 1 \mathrm{~kg}$ equals " 1.0 " --> meaning 2.2 pounds is the exact same weight as 1.0 kilogram (think of 2.2 pounds on one side of a scale and 1.0 kg on the other side -- they should be perfectly balanced). This is the key to converting units this way - whenever we multiply by a ratio of 1.0, the final answer is not affected, only the units which is what we want to change.
To say this in another way, the actual number might change (like from 10 to 22 in the example above), but the actual value of how much drug you are giving per body weight does not. We are only expressing it in a different way - in this case $\mathrm{mg} / \mathrm{kg}$ instead of $\mathrm{mg} / \mathrm{lb}$.

Here is the next step to convert to $\mathrm{g} / \mathrm{kg}$ :
$\underline{22 \mathrm{mg} x} \underline{1 \mathrm{~g}}=\underline{22 \mathrm{mg}} \times \underline{1 \mathrm{~g}}=22 / 1000=0.022 \mathrm{~g} / \mathrm{kg}$ This has given us the rest of the answer.
kg 1000mg kg 1000mg
We can also do this in one long set up:
$\frac{10 \mathrm{mg}}{\mathrm{lb}} \times \frac{2.2 \mathrm{lb}}{\mathrm{kg}} \times \frac{1 \mathrm{~g}}{1000 \mathrm{mg}}=\frac{10 \mathrm{mg}}{\mathrm{bb}} \times \frac{2.2 \mathrm{bb}}{\mathrm{kg}} \times \underset{1000 \mathrm{mg}}{\frac{1 \mathrm{~g}}{}}=0.022 \mathrm{~g} / \mathrm{kg}$ We have the same answer after cancelling lb and mg .

You can set up a long series of conversions this way - just remember to follow your units carefully!

## What happens if you set it up wrong?


This answer is clearly wrong because the units do not make sense. You want $\mathrm{g} / \mathrm{kg}$ but you got $\mathrm{g} * \mathrm{~kg} / \mathrm{lb} * \mathrm{lb}$. This means you need to look back at your setup to see what went wrong.
$10 \mathrm{mg}(\mathrm{kg} \mathrm{x}) 1 \mathrm{~g} \quad$ Here you can see that you should have inversed the circled fraction. This fixes it all!
lb $\quad 2.21 \mathrm{l}$ 1000mg

For each calculation, you should also consider if there is another way to make the same calculation. If there is, I encourage you to calculate using the alternative method. If your numbers match, then you are on the right track!

Finally, think critically about whether or not your answer makes sense. For example if you usually give a dachshund 0.2 ml of morphine, this is a very small dose for a mastiff. More dangerous is the dose of ketamine that is appropriate for a mastiff but given to a dachshund! If the answer doesn't make sense, you should double-check it. If you are ever unsure, you should not be afraid to ask a buddy to double-check you. One characteristic of a good, conscientious technician is to not be afraid to ask for assistance when needed.

## 1. Weight conversions

a. Buster weighs 17.5 kg . How many pounds does he weigh?

$$
\frac{17.5 \mathrm{~kg}}{1} \times \frac{2.2 \mathrm{lb}}{\mathrm{~kg}}=\frac{17.5 \mathrm{~kg}}{1} \times \frac{2.2 \mathrm{lb}}{\mathrm{~kg}}=17.5 \times 2.2=38.5 \mathrm{lb}
$$

b. Sasha weighs $\mathbf{1 4 . 6}$ pounds. How many kilograms does she weigh?
$\frac{14.6 \mathrm{lb}}{1} \times \frac{1}{2.2 \mathrm{lb}}=\frac{14.6 \mathrm{H}}{1} \times \frac{1}{2.2 \mathrm{Hb}}=14.6 / 2.2=6.6 \mathrm{~kg}$
c. Huxley weighs 10 pounds and 4 ounces. How many kilograms does he weigh? Sometimes you will use a scale that measures in pounds and ounces instead of a decimal value (like 2.5 pounds). In those cases, you will need to convert the total weight to pounds before you convert to other values like kilograms. There are 16 ounces in a pound, so we will need to convert ounces to a fraction (decimal value) of a pound for easier conversion.
Step 1: Convert ounces to a fraction of a pound $=\frac{40 z}{1} \times \frac{1 \mathrm{lb}}{16 \mathrm{oz}}=\frac{4 \theta z}{1} \times \frac{1 \mathrm{lb}}{16 \theta z}=4 / 16=0.25 \mathrm{lb}$

Step 2: Add the total pounds values together $=10 \mathrm{lb}+0.25 \mathrm{lb}=10.25 \mathrm{lb}$
Step 3: Convert to kilograms $=\frac{10.25 \mathrm{lb}}{1} \times \frac{1 \mathrm{~kg}}{2.2 \mathrm{lb}}=\frac{10.25 \mathrm{bb}}{1} \times \frac{1 \mathrm{~kg}}{2.2 \mathrm{Hb}}=10.25 / 2.2=4.7 \mathrm{~kg}$
${ }^{* *}$ We know these answers make sense if we think about it. A pound is the same as 2.2 kg . To say it another way, there are a little more than 2 kilograms per pound. Thinking about it this way, there is a little over twice the value of pounds compared to kilograms. We can also say there are a little less than half of the number of pounds compared to the kilogram value. Now look back at the previous problems and notice this relationship.
d. Your tiny patient, Bruno, weighs 166 grams. Your doctor is giving you drug doses in mg/kg for this patient. How many kilograms does Bruno weigh?
$166 \mathrm{~g} \times 1 \mathrm{~kg}=166 \mathrm{~g} \times 1 \mathrm{~kg}=166 / 1000=0.166 \mathrm{~kg}$
$1 \quad 1000 \mathrm{~g} 1$ 1000g
**You might notice here that we are dealing with factors of ten ( $0.001,0.01,0.1,1,10,100,1000$ ). Whenever we are dealing with multiplying or dividing by a factor of ten, we can also move the decimal place to get the same answer. For each "ten" that you are multiplying, you will move the decimal place to the right by one position. Likewise, for each "ten" you are dividing, you will move the decimal place to the left by one position. In this case, there are 1000g in each kilogram. $10 \times 10 \times 10=1000$. This is essentially three "tens" (notice the 3 zeros?). If you look at 166 g and want to convert to kilograms, we know we will get a much lower number because kilograms are a much bigger unit (so a few grams is just a small fraction of one kilogram).
First ten: $\quad 166.0 / 10$ moves the decimal one place to the left to get 16.60
Second ten: $\quad 16.60 / 10$ moves the decimal one place to the left to get 1.660
Third ten: $\quad 1.660 / 10$ moves the decimal one place to the left to get 0.1660
2. Volume and concentration calculations
a. Your doctor asks you to add 5 ml of sterile water to a 500 mg vial of ampicillin powder for injection. What is the final concentration of the injectable solution?
**In this case, we are assuming the volume of the powder is negligible

When we are asking for concentration, this generally mean we are wanting to know how much mass is in a certain volume. So in this case, we want to know how many milligrams are in each milliliter: $\underline{500 \mathrm{mg}}=500 / 5=100 \mathrm{mg} / \mathrm{ml}$
5 ml
b. Your doctor asks you to use 5 ml from a bag of $\mathbf{2 5 0 m l}$ normal ( $0.9 \%$ ) saline to reconstitute a 500 mg vial of ampicillin. Then to draw up the entire volume (all 500 mg ) of ampicillin from the bottle and add it back to the 250 ml bag of saline. What is your final concentration of ampicillin in the saline bag?

We know that all of the 500 mg are ending up in the 250 ml of saline:
$500 \mathrm{mg}=500 / 250=2.0 \mathrm{mg} / \mathrm{ml}$
250ml
c. The doctor asks you to fill 60ml of lactulose liquid medication for Claudia. You have 10z, 2oz, 3oz, and $40 z$ bottles for dispensing medication. Which one should you choose for Claudia's medication?

In this problem, we must convert milliliters to ounces. We know there are 30 ml per ounce: $\underline{60 \mathrm{ml}} \times \underline{\mathrm{oz}}=\underline{60 \mathrm{mt}} \times \underline{\mathrm{oz}}=2.0 \mathrm{oz}$ We can use a 2 oz bottle for dispensing this medication.

130 ml 130 ml
d. Daisy had an allergic reaction to a bee-sting at home. Her owners don't have a syringe but they do have measuring spoons. You would like Daisy to get 2.5 ml of Children's Elixir Benadryl. How many teaspoons should the owner give Daisy?
**Many human medications are dispensed in terms of milligrams per 5 ml . Children's Elixir is $12.5 \mathrm{mg} / 5 \mathrm{ml}$. This 5 ml unit is used because it is equivalent to 1 teaspoon which is an easy measuring device that most people have at home.
$\frac{2.5 \mathrm{ml}}{1} \times \frac{1 \mathrm{tsp}}{5 \mathrm{ml}}=\frac{2.5 \mathrm{ml}}{1} \times \frac{1 \mathrm{tsp}}{5 \mathrm{ml}}=0.5 \mathrm{tsp}$ or $1 / 2$ teaspoon
e. You have calculated the daily hydration requirement for Bernard to be about 600 ml of fluid. How many cups of water should you tell the owners is a minimum amount the dog should be drinking? We know from the conversion tables at the beginning of this document that there are 240 ml per cup. $\frac{600 \mathrm{ml}}{1} \times \frac{1 \mathrm{cup}}{240 \mathrm{ml}}=\frac{600 \mathrm{ml}}{1} \times \frac{1 \mathrm{cup}}{240 \mathrm{ml}}=600 / 240=2.5 \mathrm{cups}$
f. Your doctor asked you to draw up a $5 \mathrm{ug} / \mathrm{kg}$ dose of dexmedetomidine for Scrappy. Scrappy is 10kg and dexmedetomidine is $0.10 \mathrm{mg} / \mathrm{ml}$. How do you convert the concentration to ug/ml for easier conversion?
$\frac{0.10 \mathrm{mg}}{\mathrm{ml}} \times \frac{1000 \mathrm{ug}}{\mathrm{mg}}=\frac{0.10 \mathrm{mg}}{\mathrm{ml}} \times \frac{1000 \mathrm{ug}}{\mathrm{mg}}=0.1 \times 1000=100 \mathrm{ug} / \mathrm{ml}$
If you want to complete the first portion of this problem, you would then use the $\mathrm{ug} / \mathrm{ml}$ value to calculate a dose for Scrappy:
$5 \underline{\mathrm{ug}} \times \underline{1 \mathrm{ml}} \times \underline{10 \mathrm{~kg}}=\underline{5 \mathrm{ug}} \times \underline{1 \mathrm{ml}} \times \underline{10 \mathrm{~kg}}=0.5 \mathrm{ml}$
kg 100ug 1 kg 100ug 1
See how we oriented the fractions so that ultimately we get the units we want in the end?
g. Mannitol is $\mathbf{2 0 \%}$ solution. How many $\mathbf{~ m g} / \mathrm{ml}$ is this equivalent to?

We know from our conversion charts that \% solution is equal to the number of grams for every 100 ml of solution. In this case it is 20 g for every 100 ml
$\frac{20 \mathrm{~g}}{100 \mathrm{ml}} \times \frac{1000 \mathrm{mg}}{\mathrm{g}}=\underline{20 \mathrm{~g}} \times \underline{1000 \mathrm{mg}}=(20 \times 1000) / 100=200 \mathrm{mg} / \mathrm{ml}$
**Here's a quick trick: If you multiply \% solution by 10 , you will get the value in $\mathrm{mg} / \mathrm{ml}$. Now look at the problem above and see that this is true. Now you know why this works.
h. Your Lidocaine solution is $\mathbf{2 0 m g} / \mathrm{ml}$. What percent solution is this?

We need to convert this using grams per 100 ml as our unit of "\%" (see pervious problem)
$20 \mathrm{mg} x \underline{\mathrm{~g}}=\underline{20 \mathrm{mg}} \times \underline{\mathrm{g}}=20 / 1000=0.02 \mathrm{~g} / \mathrm{ml}$
ml 1000 mg ml 1000 mg

This gives us how many grams are in 1 ml . Now we need to know how many grams are in 100 ml - there is 100 times as much in 100 ml compared to just 1.0 ml . When we use the conversion factor of $1 \mathrm{~g} / 100 \mathrm{ml}$, then we end up with no units left and will defer to the $\%$ value in this case:
$\frac{20 \mathrm{mg}}{\mathrm{ml}} \times \frac{\mathrm{g}}{1000 \mathrm{mg}} \frac{100 \mathrm{ml}}{\mathrm{g}}=\frac{20 \mathrm{mg}}{\mathrm{ml}} \times \underset{1000 \mathrm{mg}}{\mathrm{g}} \frac{100 \mathrm{ml}}{\mathrm{g}}=0.02 \times 100=2 \%$
**Here's a quick trick: If you divide $\mathrm{mg} / \mathrm{ml}$ by 10 , you will get the value in $\%$ solution Now look at the problem above and see that this is true. Now you know why this works.
i. Your doctor has asked you to add $20 \mathrm{mEq} / \mathrm{L}$ of KCl (potassium) to a new 1 L bag of Normosol-R. KCL is $\mathbf{2 m E q} / \mathrm{ml}$. How many milliliters do you add to the bag of Norm-R?
**We will assume the volume we are adding to the bag is negligible. See the comments with dextrose calculations for more detail.
If we want $20 \mathrm{mEq} / \mathrm{L}$, then this problem is easy because we just have to figure out how many ml of KCl it takes to get 20 mEq . This is because we are adding to 1 L .
$\frac{20 \mathrm{mEq}}{1} \times \frac{1 \mathrm{ml}}{2 \mathrm{mEq}}=\frac{20 \mathrm{mEq}}{1} \times \underline{2 \mathrm{ml}}=20 / 2=10 \mathrm{ml}$
j. Your doctor has asked you to add 20mEq of KCl (potassium) to a 1 L bag of Normosol-R that right now only has 600 ml left after running 400 ml into the patient already. KCL is $2 \mathrm{mEq} / \mathrm{ml}$. How many milliliters do you add to the 600 ml bag of Norm-R?
**This problem is a little trickier. I like to think in terms of ratios. So if there is $600 / 1000 \mathrm{ml}$, this means there is $60 \% ~(600 / 1000 \times 100 \%=60 \%)$ of the fluid left in the bag. This means we only need $60 \%$ of whatever you would have calculated for 1.0L. I find it's easier to calculate how much you would need for one liter, then determine what $60 \%$ of that value would be:

We use the previous calculation for determining how many milliliters we need of KCl for 1.0 L of Norm-R: $\underline{20 \mathrm{mEq}} \times \underline{1 \mathrm{ml}}=\underline{20 \mathrm{mEq} \times \underline{1 \mathrm{ml}}=20 / 2=10 \mathrm{ml}, ~}$

$$
1 \quad 2 \mathrm{mEq} \quad 1 \quad 2 \mathrm{mEq}
$$

$10 \mathrm{ml} \times 60 \%=10 \mathrm{ml} \times 0.6=6 \mathrm{ml}$
**If you are ever not confident about your answer, you can always find a way to double check yourself. In this case you will take your 6 ml of KCl , determine how many mEq are in that 6 ml , then calculate your total $\mathrm{mEq} / 600 \mathrm{ml}$, then finally convert milliliters to liters (each of these steps is sequentially reflected below in the series of multiplied fractions):
$\underline{6 \mathrm{ml}} \times \underline{2 \mathrm{mEq}} \times \underline{1} \quad \underline{1000 \mathrm{ml}}=\underline{6 \mathrm{ml}} \times \underline{2 \mathrm{mEq}} \times \underline{1} \quad \underline{1000 \mathrm{ml}}=20 \mathrm{mEq} / \mathrm{L}$
1 ml 600 ml 1L 1 ml 600ml 1L
k. Your doctor has asked you make up a 1L bag of LRS with 5\% dextrose. The dextrose is 50\% concentration. How do you prepare the fluid bag?
This is a situation in which you can use the following equation:
$\mathrm{C} 1 \mathrm{~V} 1=\mathrm{C} 2 \mathrm{~V} 2$
$\mathrm{C} 1=$ concentration of the Fluid A
$C 2=$ concentration of Fluid $B$
V1=volume of Fluid $A$
V2=final volume of Fluid $B$

Anytime you have one fluid of a particular concentration that you are adding to another fluid, you can use this formula.

In this situation:
Fluid $A=$ dextrose $50 \%$ solution
Fluid $B=5 \%$ LRS fluid solution
$C 1=$ concentration of the Fluid $A=50 \%$
$C 2=$ concentration of Fluid $B=5 \%$
$\mathrm{V} 1=$ volume of Fluid $\mathrm{A}=$ unknown - this is what we are trying to determine so we will call it V1
$\mathrm{V} 2=$ final volume of Fluid $B=1000 \mathrm{ml}$
$(50 \%)(\mathrm{V} 1)=(5 \%)(1000 \mathrm{ml})-->(50 \%) \mathrm{V} 1=5000 \% * \mathrm{ml}-->\underset{50 \%}{5000 \% * \mathrm{ml}}=\mathrm{V} 1-->\mathrm{V} 1=100 \mathrm{ml}$ of $50 \%$ dextrose
**In this case, 100 ml seems like a significantly large volume compared to the 1000 ml in the bag. This math problem assumes your final volume is 1000 ml , however if you simply add 100 ml of $50 \%$ dextrose to the LRS, you end up with the volume of LRS plus the dextrose which is $1000+100=1100$. To be more accurate, you should:
First remove 100 ml of plain LRS from the bag-discard
Next add 100 ml of $50 \%$ dextrose to the 900 ml of LRS
Now you have 1000 ml of $5 \%$ dextrose in your final mixture
**To be precise, this should be done anytime you are adding fluid volumes (like our injectable metoclopramide), however those volumes are usually very small and considered negligible compared to the larger volumes needed to make a dextrose CRI. Even for this problem, the difference when not taking this step is not large. 100 ml of $50 \%$ dextrose to end up with 1100 ml gets you about $4.5 \%$ dextrose. Likewise, 50 ml of $50 \%$ dextrose to end up with 1050 ml is about $2.4 \%$ which is not far from the intended $2.5 \%$. I bring this up here, because as you get to larger proportions of fluid volumes that you are adding, the volume will become more significant and you must understand how this relationship is supposed to work and then decide if you must remove volume (as described above) in your preparation to keep it more accurate. If you are ever unsure, ask your doctor what he/she prefers.
I. Your doctor has asked you add dextrose to a patient's LRS fluid bag to make a 2.5\% dextrose solution. The dextrose is $50 \%$ concentration and the LRS fluid bag only has 600 ml left out of the original 1000 ml . How do you prepare the fluid bag?
This is similar to the previous problem --> in this case:
Fluid $A=$ dextrose $50 \%$ solution
Fluid $B=2.5 \%$ LRS fluid solution
$C 1=$ concentration of the Fluid $A=50 \%$
$C 2=$ concentration of Fluid $B=2.5 \%$
$\mathrm{V} 1=$ volume of Fluid $\mathrm{A}=$ unknown - this is what we are trying to determine so we will call it V1
$\mathrm{V} 2=$ final volume of Fluid $B=600 \mathrm{ml}$

$$
(50 \%)(\mathrm{V} 1)=(2.5 \%)(600 \mathrm{ml}) ~-->(50 \%) \mathrm{V} 1=1500 \% * \mathrm{ml}-->\frac{1500 \% * \mathrm{ml}}{50 \%}=\mathrm{V} 1 \quad->\mathrm{V} 1=30 \mathrm{ml} \text { of } 50 \% \text { dextrose }
$$

**In this case, you can remove 30 ml of the original LRS solution and discard Then you will add 30 ml of $50 \%$ dextrose to give you the final 600 ml of $2.5 \%$ dextrose solution. Alternatively if you leave it as 30 ml plus 600 ml for a final concentration of $2.4 \%$.

For the following problems, I am not going to elaborate each step as I did for the previous problems. By now you should be getting the hang of it, so I will focus on elaborating more unusual or tricky steps

## 3. Filling prescriptions

a. Your doctor has asked you to fill 4 week's worth of $1 / 4$ grain ( 16.2 mg ) Phenobarbital tablets for Sassy. She gets $\mathbf{1 / 2}$ of a $\mathbf{1 / 4}$ grain tablet every 12 hours. How many tablets should you fill for Sassy? First we need to figure out how many tablets Sassy gets every day. If Sassy gets $1 / 2$ of a tablet twice a day, then she gets 1 tablet per day.

dose day
We then multiply the number of tablets she gets per day by the number of days she needs the medication to get the total number needed. In this case it is 1 tablet multiplied by 28 days (because there are 7 days a week and 4 weeks total).
$\underline{1 \text { tablet } \times 4 \text { weeks } \times \underline{7 \text { days }}=1 \times 4 \times 7=28 \text { tablets } .}$

```
day 1 week
```

b. Your doctor has asked you to fill 3 weeks of gabapentin for Elsworth. She should get about $10 \mathrm{mg} / \mathrm{kg}$ of gabapentin every 8 hours. She weighs 9.2 kg . Your clinic carries $50 \mathrm{mg}, 100 \mathrm{mg}$, and 300 mg capsule sizes. Which size do you choose and how many do you fill for Elsworth?
First we need to figure out what size capsule she needs to take:
$9.2 \mathrm{~kg} \times 10 \mathrm{mg}=92 \mathrm{mg}$
$1 \quad \mathrm{~kg}$
This is very close to the 100 mg capsule size, so this is what we will choose. Next we need to decide how many capsules she needs in total. We will do what we did in the previous problem by figuring out how many capsules she needs per day, then how many total for the 3 weeks. We will put this together in one long problem below:
$\frac{1.0 \text { capsules }}{\text { dose }} \times \frac{3 \text { doses }}{\text { day }} \times \frac{3}{1} \times \frac{7 \text { days }}{}=1 \times 3 \times 3 \times 7=63$ capsules
c. Your doctor has prescribed 0.8 ml of $50 \mathrm{mg} / \mathrm{ml}$ oral suspension gabapentin every $\mathbf{8}$ hours for $\mathbf{2}$ weeks for Sparky. How many milliliters of the gabapentin suspension do you need to fill for Sparky? We treat this the same way we do for pills and capsules. We first figure out how much he needs per day, then we multiply by the number of days he needs the medication:
**In cases of liquids, I will often fill a little bit more to account for losses in the syringe hub or if the owner might need to redose. You might ask your doctor if they want to round up to 35 ml for these reasons.
d. Your doctor has asked you to make 10 syringes of 0.3 ml of buprenorphine for Fluffernutter. How many milliliters will you charge for in total?
We need to multiply the volume by the number of syringes to get the total volume you are drawing up:
$0.3 \mathrm{ml} \times 10$ syringes $=3.0 \mathrm{ml}$
syringe
**Remember how we can move the decimal point over to the right one space for each ten we use to multiply? See how all we need to do is to just move the decimal point here because we are multiplying by ten?

## 4. Dosing injections

a. You have been asked to give Juno $0.3 \mathrm{mg} / \mathrm{kg}$ of methadone IV for pain control. Juno weighs $\mathbf{2 4}$ pounds. Methadone is $10 \mathrm{mg} / \mathrm{ml}$. How many milliliters will you give to Juno?
First we must convert Juno's weight to kilograms:
$\underline{24 \mathrm{lb}} \times \underline{1 \mathrm{~kg}}=24 / 2.2=10.9 \mathrm{~kg}$
12.2 H

Next we must find the volume of methadone required:
$10.9 \mathrm{~kg} \times \underline{0.3 \mathrm{mg}} \times \underline{1 \mathrm{ml}}=(10.9 \times 0.3) / 10=0.3 \mathrm{ml}$
$1 \quad \mathrm{~kg} \quad 10 \mathrm{mg}$
b. Your doctor has asked you to give $\mathbf{0 . 5}$ grams/kg of mannitol IV over $\mathbf{2 0}$ minutes to Charlie. Mannitol is $\mathbf{2 0 \%}$ solution and Charlie weighs 17.4 kg . How many milliliters will you give?
**Remember that \% solution is the same as the number of grams per 100ml? So $20 \%$ is the same as $20 \mathrm{~g} / 100 \mathrm{ml}$
$\frac{17.4 \mathrm{~kg}}{1} \times \frac{0.5 \mathrm{~g}}{\mathrm{~kg}} \times \frac{100 \mathrm{ml}}{20 \mathrm{~g}}=(17.4 \times 0.5 \times 100) / 20=43.5 \mathrm{ml}$
c. Your doctor needs you to give 10 units (U) of Humulin R insulin to Juliette, but you do not have U-100 insulin syringes. Humulin $R$ is $100 \mathrm{U} / \mathrm{ml}$. How many milliliters will you draw up to use and which size syringe ( $1 \mathrm{ml}, 3 \mathrm{ml}, 6 \mathrm{ml}$ ) syringe should you use for this?
$\underline{10 U} \times 1 \mathrm{ml}=10 / 100=0.1 \mathrm{ml}$
1 100 $\forall$
This is best given with the smallest syringe you can so you can be as accurate as possible. You should use a 1 ml syringe here.
d. Your doctor needs you to give 10 units (U) of Vetsulin insulin to Juliette, but you do not have U-40 insulin syringes. Vetsulin is $40 \mathrm{U} / \mathrm{ml}$. How many milliliters will you draw up to use and which size syringe ( $1 \mathrm{ml}, 3 \mathrm{ml}, 6 \mathrm{ml}$ ) syringe should you use for this?
$\frac{10 \mathrm{U}}{1} \times \frac{1 \mathrm{ml}}{40 \mathrm{U}}=10 / 40=0.25$
This is best given with the smallest syringe you can so you can be as accurate as possible. You should use a 1 ml syringe here.
e. Challenging question: Your client picked up her new prescription for Vetsulin for her dog Rexie who is getting 8 units of Vetsulin insulin SQ twice a day. Vetsulin is $40 \mathrm{U} / \mathrm{ml}$ and requires $\mathrm{U}-\mathbf{4 0}$ syringes. The client was accidentally sent home with $0.5 \mathrm{ml} \mathrm{U}-100$ syringes and right now the clinic is closed. Can you tell her how to give the evening and morning doses using the $\mathrm{U}-100$ syringes? How many units using the $\mathbf{U - 1 0 0}$ syringes is the equivalent of $\mathbf{8 U}$ of $\mathbf{4 0 U} / \mathrm{ml}$ Vetsulin?
First we need to figure out what volume of insulin Rexie needs:
$\underline{8 甘} \times \underline{1 \mathrm{ml}}=8 / 40=0.2 \mathrm{ml}$
$140 U$

Next we need to figure out how to measure volume using the U-100 syringes. We know each U-100 syringe holds 0.5 ml . This half-milliliter is divided into 50 units (this is because the syringe is 0.5 ml , so we will need to be able to measure half of the units that are in each milliliter. If it is 1000 for each milliliter, then the 0.5 ml syringe holds half of this or 50 units.

There are 50 units for the 0.5 ml and we want to know specifically how many milliliters per unit is represented
$\underline{0.5 \mathrm{ml}}=0.01 \mathrm{ml} / \mathrm{U}$
50 U
So each unit mark on the syringe represents 0.01 ml . We know from our first calculation that we need 0.2 ml . Next we need to figure out how many units will give us 0.2 ml :
$\underline{0.2 \mathrm{ml}} \times \underline{1 \mathrm{U}}=20 \mathrm{U}$ So you will draw up 20 U on the U100 syringe to equal 8 U of $40 \mathrm{U} / \mathrm{ml}$ insulin
$1 \quad 0.01 \mathrm{ml}$
**We know this answer can make sense because regular insulin is 2.5 times more concentrated that Vetsulin $(100 / 40=2.5)$. This means it takes 2.5 times more volume to give the $40 U / \mathrm{ml}$ insulin compared to $100 \mathrm{U} / \mathrm{ml}$. Do you see that 20 U is 2.5 times greater than 8 U ?

## 5. Calculating fluid rates

a. Your doctor has asked you to set up LRS fluids at a maintenance rate for a dog named Jackson. She would like you to use $60 \mathrm{ml} / \mathrm{kg} /$ day as the maintenance rate. Jackson is 60 pounds. What fluid rate will you use for the LRS?
First, we must convert Jackson's weight to kilograms:
$\underline{60 \mathrm{Hb}} \times \underline{1 \mathrm{~kg}}=60 / 2.2=27.3 \mathrm{~kg}$
12.2 b

Next, we must determine how many milliliters Jackson will be getting each day:
$\underline{60 \mathrm{ml}} \times \underline{27.3 \mathrm{~kg}}=60 \times 27.3=1638 \mathrm{ml} /$ day
kg*day 1

Finally, we must determine how many milliliters per hour to set the fluid pump. There are 24 hours in a day, so we divide the total daily fluid amount by 24 hours:
$1638 \mathrm{ml} \times 1$ day $=1638 / 24=68.25$ (round to $68 \mathrm{ml} / \mathrm{hour}$ )
day 24 hours

This can all be set up in one long series of calculations:
$\underline{60 \mathrm{~b}} \times \underline{1 \mathrm{~kg}} \times \underline{60 \mathrm{ml}} \times \underline{1 \text { day }}=(60 \times 60) /(2.2 \times 24)=68.18$ (round to $68 \mathrm{ml} /$ hour)
$12.2 \mathrm{lb} \mathrm{kg}^{*}$ day 24 hours
**The second answer is 68.18 instead of 68.25 because in the first answer I rounded answers for each step - this can slightly affect your final calculations requiring a longer series of steps. In the second problem, I did all the calculations at once so this is a more precise number.
**Did you notice that the maintenance rate for this dog is close to the same value as his weight in pounds? This can give you a quick estimate of what to expect the correct answer should be. This is for a DOG. Cats have a lower daily maintenance rate of $45 \mathrm{ml} / \mathrm{kg} / \mathrm{hour}$. Can you estimate the hourly rate for a 10-pound cat based on this information?
Consider that if you had a dog, the estimated rate would be $10 \mathrm{ml} /$ hour. However a cat's rate is 3/4 (same as $75 \%$ ) of that of a dog (because 45 is $75 \%$ of 60 ). In that case, you would take $75 \%$ of 10: $10 \mathrm{ml} /$ hour x $0.75=7.5 \mathrm{ml} /$ hour .
b. Your doctor has asked you to set up LRS fluids at a maintenance rate for a cat named Percy. She would like you to use $45 \mathrm{ml} / \mathrm{kg} /$ day as the maintenance rate. Percy is 10 pounds and in good body condition. What fluid rate will you use for the LRS?
$\underline{10 \mathrm{Hb}} \times \underline{1 \mathrm{~kg}} \times \underline{45 \mathrm{ml}} \times \underline{1 \text { day }}=(10 \times 45) /(2.2 \times 24)=8.5 \mathrm{ml} /$ hour
$12.2 \mathrm{Hb} \mathrm{kg}^{*}$ day 24 hours
**Refer to the previous question that addresses quick estimates for hourly maintenance fluid rates. Do you see how this answer is close to the estimated rate of $7.5 \mathrm{ml} / \mathrm{hour}$ ( $75 \%$ of 10lb)? This tells you that you have calculated correctly.
c. Your doctor has asked you to set up LRS fluids at a maintenance rate for a cat named Toby. She would like you to use $45 \mathrm{ml} / \mathrm{kg} /$ day as the maintenance rate. Toby is 18 pounds and morbidly obese - you estimate that he may weigh 12 pounds in lean body mass. What fluid rate will you use for the LRS? Obese cats can be easy to fluid overload. For this reason, it can be prudent to ask your doctor if a calculation based on his ACTUAL weight or estimated LEAN weight should be used. I will usually err on the side of using the estimated lean weight in an otherwise hydrated cat:
$\underline{12 \mathrm{H}} \times \underline{1 \mathrm{~kg}} \times 45 \mathrm{ml} \times 1$ day $=(12 \times 45) /(2.2 \times 24)=10 \mathrm{ml} /$ hour
$12.2 \mathrm{Hb} \mathrm{kg}^{*}$ day 24 hours
d. Your doctor has asked you to give a 200 ml bolus of LRS fluids over 15 minutes. What fluid rate in $\mathrm{ml} /$ hour should you set the fluid pump to?
We know how much volume we want to give per 15 minutes, now we need to convert to how much fluid we want to give in one hour:
$200 \mathrm{ml} \times 60$ minutes $=(200 \times 60) / 15=800 \mathrm{ml} /$ hour
$15 \mathrm{~min} \quad 1$ hour
**Do you see how 800 is 4 times as much as 200? Do you also appreciate that 1 hour is 4 times longer than 15 minutes (think of a clock and how the 15 -minute position is $1 / 4$ of the entire circle). A quick way to think about this is to just multiply the fluid volume by 4 here. You are not really giving 800 ml because you are stopping $1 / 4$ of the way into the hour $-1 / 4$ of 800 is the 200 ml you want to give. Does this make sense?
e. Your doctor has asked you to give 400 ml of Norm-R fluids over $\mathbf{2 0}$ minutes. What fluid rate in $\mathbf{~ m l} /$ hour should you set the fluid pump to?
This is very similar to the last problem only we are now using 20 minutes instead of 15:
$400 \mathrm{ml} \times 60$ minutes $=(400 \times 60) / 20=1200 \mathrm{ml} /$ hour
20 min 1 hour
**Now there are only 3 segments of 20 minutes in each hour. In this case, you can multiply 400 ml by 3 to get 1200 as your quick calculation. In order to get 400 ml into your patient in $1 / 3$ of an hour, you have to triple your hourly fluid rate.

## 6. Calculating CRIs

a. You are setting up a fentanyl CRI for surgery. The patient weighs 26 kg , the fentanyl is $50 \mathrm{ug} / \mathrm{ml}$, and you want to have $4 \mathrm{ug} / \mathrm{kg} / \mathrm{hour}$ rate. You need to have enough fentanyl at this rate to run for 3 hours. What is the rate in $\mathrm{ml} /$ hour? How much fentanyl will you need to have to last all 3 hours ?
$\underline{26 \mathrm{~kg}} \times \underline{4 \mathrm{ug}} \times \underline{1 \mathrm{ml}}=(26 \times 4) / 50=2.1 \mathrm{ml} /$ hour
1 kg*hr 50ug
Do you see how we combined two steps in the above calculation? First we determined how many micrograms would be used per hour. Then we converted to milliliters per hour.

Now we must determine how many milliliters we need for 3 hours:
$2.1 \mathrm{ml} \times 3$ hours $=6.3 \mathrm{ml}$ will be used in 3 hours
hour 1
**We must also remember that you will likely have an infusion set up using IV tubing. You will need to take the volume of this infusion set-up into your calculation. So if it takes 3.0 ml to fill the infusion line to the patient, you will also need to add this to the total volume needed:
$6.3 \mathrm{ml}+3.0 \mathrm{ml}=9.3 \mathrm{ml}$ total
This volume can vary depending on the equipment you use. Ask your doctor or a fellow technician if you are unsure of the volume you should add to your estimate.
b. You need to prepare propofol for induction and maintenance of anesthesia for Louie. Louie is 42kg and propofol is $10 \mathrm{mg} / \mathrm{ml}$. You will need to plan for up to $4 \mathrm{mg} / \mathrm{kg}$ to use for induction. Then you will use $0.4 \mathrm{mg} / \mathrm{kg} / \mathrm{min}$ to maintain anesthesia. How much volume do you need to have ready for induction? What rate of propofol will you use for Louie in $\mathrm{ml} / \mathrm{min}$ ? What is the rate in $\mathrm{ml} / \mathrm{hour}$ ? How much volume should you have available for 45 minutes of anesthesia? $42 \mathrm{~kg} \times \underline{4 \mathrm{mg}} \times 1 \mathrm{ml}=(42 \times 4) / 10=16.8$ rounded up to 17 ml available to induce
1 kg 10 mg
$\underline{42 \mathrm{~kg}} \times \underline{0.4 \mathrm{mg}} \times \underline{1 \mathrm{ml}}=(42 \times 0.4) / 10=1.7 \mathrm{ml} / \mathrm{min}$ CRI rate
$1 \mathrm{~kg} * \min 10 \mathrm{mg}$
$1.7 \mathrm{ml} \times \underline{60 \mathrm{~min}}=102 \mathrm{ml} /$ hour CRI rate See how this was a simple conversion from minutes to hours? min hour
$1.7 \mathrm{ml} \times 45 \mathrm{~min}=76.5 \mathrm{ml}$ used in 45 minutes
min 1
** Remember you must add the volume of your fluid infusion lines to the calculated volume required. So if you have a long fluid line that holds 10 ml , then you must add 10 ml to the calculated 76.5 ml for a total of 86.5 ml .
c. Xander needs to get $1.5 \mathrm{mg} / \mathrm{kg} /$ day of metoclopramide as a CRI. Your doctor would like you to add this to Xander's IV fluids. Xander is 15 kg and metoclopramide is $5 \mathrm{mg} / \mathrm{kg}$. His fluid rate is $\mathbf{3 5 m l} / \mathrm{hour}$. How many milliliters of metoclopramide to you need to add to his 1 liter bag of LRS?
First we must determine how many milligrams of metoclopramide Xander needs per day:
$15 \mathrm{~kg} \times 1.5 \mathrm{mg}=22.5 \mathrm{mg}$
1 kg*day

Next we must determine how much fluid he will get each day:
$\underline{35 \mathrm{ml}} \times \underline{24}$ hours $=840 \mathrm{ml} /$ day
hour day

Now we see that Xander will use less than 1 liter per day, so we already know we will need more than 22.5 mg in the 1 L fluid bag. We can make a proportion and cross multiply here. This is something we have not specifically showed you before. It is based on the idea that you can set up equal ratios that allow you to solve for a missing number. For example:
$\underline{1}=\underline{3}$
412

We know that $1 / 4$ and $3 / 12$ are both the same number ( 0.25 ). What if we didn't know what was 0.25 of 12?
$\underline{1}=\underline{x}$
412

Mathematical principles allows us to multiply diagonally across the equal sign:
$\underline{1}=\underline{x} \quad 4 x=12$
4

Now to solve for "x", we will need to divide 12 by 4 (this assumes you know rules of basic algebra). In this case $12 / 4=x=3$

We use this principle of cross-multiplication to figure out which ratio will be equal to the daily requirement of 22.5 mg in 840 ml :
 840 ml 1000ml

$$
x=22,500 \mathrm{mg}^{*} \mathrm{mt} / 840 \mathrm{mt}=26.8 \mathrm{mg}
$$

So now we know that 26.8 mg in 1000 ml (1L) of fluids gives us the same proportion as 22.5 mg in 840 ml . We therefore need to add 26.8 mg to 1 L of LRS. We know the concentration of metoclopramide is $5 \mathrm{mg} / \mathrm{ml}$ so we know need to determine how many milliliters to add:

15 mg
d. Xander needs to get $1.5 \mathrm{mg} / \mathrm{kg} /$ day of metoclopramide as a CRI. Your doctor would like you to add this to Xander's IV fluids. Xander is 15 kg and metoclopramide is $5 \mathrm{mg} / \mathrm{kg}$. His fluid rate is $35 \mathrm{ml} / \mathrm{hour}$. How many milliliters of metoclopramide to you need to add to the bag of LRS fluids? Right now he has 750 ml left in his 1 L bag of LRS.
This is going to be similar to the problem in which you calculated how much KCl to add to a partial fluid bag. The first step here is to determine how many milliliters of metoclopramide to add to 1 L of LRS. You did this in the previous question and came up with 5.4 ml in 1 L . Now you need to determine how many milliliters you will use for 750 ml :
$750 / 1000=75 \%$ or otherwise stated as 0.75
You will therefore only need $75 \%$ of the amount of metoclopramide you would have used for one full liter:
$5.4 \mathrm{ml} \times 0.75=4.1 \mathrm{ml}$ needed to add to 750 ml to run at $1.5 \mathrm{mg} / \mathrm{kg} /$ day at a fluid rate of $35 \mathrm{ml} / \mathrm{hour}$

## 7. ER drug quick tricks

a. Emergency doses of diazepam, atropine, and epinephrine can all be estimated the following way: 1 ml per every 20 pounds or 0.5 ml per cat-sized dog

Glance quickly at your patient and consider how many 20 pound dogs will fit (like 4 beagles into a big Labrador). Then multiply that by 1 ml .
If your patient is a big lab that looks like the size for 4 beagles, you will get 4 ml of the drug you need.
Let's look at diazepam for an 80lb dog:
Diazepam IV dose for seizures is $0.5 \mathrm{mg} / \mathrm{kg}$ and diazepam is $5 \mathrm{mg} / \mathrm{ml}$
$\underline{801 b} \times \underline{1 \mathrm{~kg}} \times \underline{0.5 \mathrm{mg}} \times \underline{1 \mathrm{ml}}=3.6 \mathrm{ml} \quad 4.0 \mathrm{ml}=0.55 \mathrm{mg} / \mathrm{kg}$--> pretty darn close to $0.5 \mathrm{mg} / \mathrm{kg}$ !
$1 \quad 2.2 \mathrm{Hb} \quad \mathrm{kg} \quad 5 \mathrm{mg}$

Let's look at atropine for an 80lb dog:
Atropine IV dose for CPR is $0.05 \mathrm{mg} / \mathrm{kg}$ (range $0.04-.01$ ) and atropine is $0.54 \mathrm{mg} / \mathrm{ml}$
$\frac{80 \mathrm{lb}}{1} \times \frac{1 \mathrm{~kg}}{22 \mathrm{lb}} \times \frac{0.05 \mathrm{mg}}{\mathrm{kg}} \times \underline{1 \mathrm{ml}}=3.4 \mathrm{ml} \quad 4.0 \mathrm{ml}=0.06 \mathrm{mg} / \mathrm{kg}$--> pretty darn close to $0.05 \mathrm{mg} / \mathrm{kg}$ !
$1 \quad 2.2 \mathrm{lb} \quad \mathrm{kg} \quad 0.54 \mathrm{mg}$

Let's look at epinephrine for an 80lb dog:
Epinephrine IV high dose for CPR follows this short cut. These days, it is suggested to start with a low dose first which is $1 / 10$ th the volume of the high dose. You can still use this shortcut to find the high dose volume, then give $1 / 10$ th of this volume to start.
Epinephrine low dose is $0.01 \mathrm{mg} / \mathrm{kg}$ (range 0.01-0.1(high dose)) and epinephrine is $1.0 \mathrm{mg} / \mathrm{ml}$ $\underline{80 \mathrm{lb}} \times 1 \mathrm{~kg} \times \underline{0.01 \mathrm{mg}} \times \underline{1 \mathrm{ml}}=0.36 \mathrm{ml}$ for the low dose, 3.6 ml for the high dose
$12.2 \mathrm{lb} \mathrm{kg} \quad 1.0 \mathrm{mg}$

$$
4.0 \mathrm{ml}=0.11 \mathrm{mg} / \mathrm{kg}-\text { target dose for the high dose epinephrine! }
$$

**Now you see how closely these doses are to the calculated doses!
b. Shock doses of crystalloid fluids

The shock dose of crystalloid fluids for a dog is about $90 \mathrm{ml} / \mathrm{kg}$. We like to give this in "quarter shock dose boluses" meaning we only give $1 / 4$ of the total estimated fluid volume at a time to make sure we don't run the risk of fluid overload. Let's use a 20 pound dog as an example:
$\underline{20 \mathrm{lb}} \times \underline{1 \mathrm{~kg}} \times \underline{90 \mathrm{ml}} \times \underline{1}=204 \mathrm{ml}$ for each quarter dose
12.2 lb kg 4

Now look at the weight in pounds. See that it is 20? Now add a zero to the end of this to get the number 200. See how 200 is nearly the same as 204 ml that took much longer to calculate? It is much quicker to add a zero to the end of a dog's weight in pounds to give you the quarter shock dose to use for fluid resuscitation.

A cat's shock dose is $45 \mathrm{ml} / \mathrm{kg}$. This is HALF of the dog dose of $90 \mathrm{ml} / \mathrm{kg}$. This means you can give a cat HALF of what you would give a dog. So if you had a 10 pound cat, you would use 100 ml if it was a dog. But since it's a cat, you will use half of this and use 50 ml instead. Does this make sense?

## c. Tips and tricks for anesthesia

Anesthesia can be scary and requires split second decisions and quick calculations. As you have seen in this document, some calculations take time. For this reason, it is smart to be prepared with precalculated doses for medications you might be more likely to use like atropine, glycopyrrolate, fentanyl and propofol boluses, etc. Also keep a chart of volume rate changes for medication CRI adjustments.

Remember to always check your work (especially your units!), always ask questions when you are not absolutely sure, and keep up the good work!

